XANTHONES FROM TOVOMITA EXCELSA*

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Abstract—The trunk wood of *Tovomita excelsa* (Guttiferae) from north-eastern Brazil contains, besides betulinic acid, 1,5-dihydroxy-6-methoxyxanthone, 1,6-dihydroxy-5-methoxyxanthone, 5,6-dihydroxy-1-methoxyxanthone and 1,7-dihydroxy-6-methoxyxanthone

The five Brazilian Tovomita species (Guttiferae), T brasiliensis [2], T choisyana [3], T macrophylla [4], T mangle [5] and T pyrifolium [6], were shown to contain sitosterol, stigmasterol, β-amyrin and betulinic acid, besides eight xanthones differing in oxygenation and prenylation patterns. The present report concerns a fifth species, T excelsa Andrade-Lima et G Mariz [7]. Its wood yielded sitostenone, sitosterol, stigmasterol, heptacosanoic acid and betulinic acid, besides the four simple xanthones 1a, 1b, 1c and 2 1,5-Dihydroxy-6-methoxyxanthone (1a) has been prepared synthetically and buchanaxanthone (1b) has been isolated from another Guttiferae species, Garcinia buchananii [8], these were identified by spectral comparisons. The other two xanthones have not been described previously

Mass spectra suggested both compounds to be dihydroxymonomethoxyxanthones The ¹H NMR spectrum of compound 1c in CDCl₃ revealed the presence of two groups of adjacent aromatic protons One group with two *ortho* protons must be placed at the 7,8-positions Indeed the low field signal $(\delta 797)$ of one of the two doublets (J = 9 Hz) is compatible only with the *peri*position of the corresponding hydrogen The other group

with two ortho, meta (broad doublets, J=85 Hz) and one ortho, ortho (t, J=85 Hz) coupled protons must be placed at the 2,3,4-positions Indeed both broad doublets ($\delta 682$ and 697) appear at a relatively high field incompatible with their location at C-1 This peri-position should be occupied by the methoxyl in view of the chemical shift of its signal ($\delta 414$) which denotes strong deshielding. The two hydroxyls must consequently occupy the 5,6-positions, their vicinality being assured by the lability of the compound in alkali and the modification of its UV spectrum upon addition of H_3BO_3 and NaOAc

The other novel xanthone (2) was soluble enough for $^1\text{H NMR}$ spectroscopy in CDCl₃ only after acetylation. The spectrum of the diacetate revealed the presence of two para related protons (two singlets, δ 6 93 and 7 88), which hence can only occupy the 5,8-positions, and three adjacent protons (2 dd, J=85 and 1 Hz, 1t, J=85 Hz). The doublets appear at relatively high field (δ 7 00, 7 37) and C-1 must thus be substituted. Indeed the compound possesses a chelatable hydroxyl at this position, as indicated by an AlCl₃ UV shift. The second hydroxyl cannot be situated at C-6 where if would impart strong acidity, incompatible with the negligible NaOAc UV shift of 2

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EXPERIMENTAL

Isolation of the constituents Trunk wood of T excelsa was collected in a forest region near São Miguel dos Campos,

$$R^{3}O$$
 OR^{2}

1a R¹= R² = H, R³ = Me 1b R¹= R³ = H, R² = Me 1c R¹= Me, R² = R³ = H

2

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Alagoas, Brazil [7], reduced to powder (3 1 kg) and percolated with C_6H_6 . The extract (15 g) was chromatographed on a silica gel column (280 g). The following fractions of 100 ml were eluted in order with the indicated solvents. Frs 1–8 solid (19 mg, C_6H_6), 9–27 sitostenone + sitosterol (32 mg), C_6H_6 —CHCl₃, 20 1), 28–79 sitosterol + stigmasterol (64 mg, C_6H_6 —CHCl₃, 5 1), 80–120· 1a (13 mg, C_6H_6 —CHCl₃, 1 3), 121–137 1c (7 mg, C_6H_6 —CHCl₃, 1 5), 138–150· 1b (16 mg, C_6H_6 —CHCl₃, 1 7), 151–160· 2 (30 mg, CHCl₃), 161–198 heptacosanoic acid (18 mg, CHCl₃—MeOH, 99 1), 199–219 betulinic acid (260 mg, CHCl₃—MeOH, 97 3)

5,6-Dihydroxy-1-methoxyxanthone (1c) Mp 218–223° (washed with MeOH) [M]⁺ found 258 0538, $C_{14}H_{10}O_5$ requires 258 0528 MS m/z (rel int) 259 [M+1] (9), 258 [M]⁺ (100), 243 (51), 215 (34), 187 (27) IR v_{max}^{KBr} cm⁻¹ 3380, 1648, 1608, 1580, 1520, 1485, 1460, 1365, 1335, 1288, 1238, 1202, 1165, 1078, 1065, 1025, 805, 778, 748, 694 UV λ_{max}^{EIOH} nm 238, 243, 265 sh, 290 sh, 375 (ε 21800, 21900, 11900, 5800, 12600), $\lambda_{max}^{EIOH+NaOH}$ nm 238, 248, 265, 285 sh, 350 sh, 385 (ε 22300, 22700, 13900, 5800, 8800, 11600), $\lambda_{max}^{EIOH+NaOAc}$ nm 235, 243, 265, 285 sh, 373 (ε 21700, 21400, 13200, 5800, 15100), $\lambda_{max}^{EIOH+NaOAc+H_3BO_3}$ nm 235, 246, 290, 325, 368 (ε 21600, 22000, 5300, 6100, 9500) ¹H NMR (CDCl₃, 270 MHz) δ 7 97 (d, d = 9 Hz, H-8), 7 03 (d, d = 9 Hz, H-7), 6 97 (dr, dr = 8 5 Hz, H-4), 7 58 (tr, dr = 8 5 Hz, H-3), 6 82 (dr dr = 8 5 Hz, H-2), 4 14 (dr (s) OMe)

1,7-Dihydroxy-6-methoxyxanthone (2) Mp 265–268° (washed with CHCl₃) [M]⁺ found 258 0520, $C_{14}H_{10}O_5$ requires 258 0528 MS m/z (rel int) 259 [M+1] (13), 258 (100), 243 (13), 215 (22), 187 (21) IR $v_{\rm MBT}^{\rm KBT}$ cm⁻¹ 3450, 1647, 1628, 1600, 1578, 1493, 1465, 1436, 1400, 1368, 1342, 1298, 1275, 1234, 1205, 1160, 1050, 1010, 848, 815, UV $\lambda_{\rm max}^{\rm MeOH}$ nm 253, 272 sh, 293, 375 (\$\varepsilon 24000, 12600, 10100, 8100), $\lambda_{\rm max}^{\rm MeOH+NaOH}$ nm 262, 280 sh (\$\varepsilon 22500, 11100), $\lambda_{\rm max}^{\rm MeOH+NaOAc}$ nm 252, 272 inf, 293, 377 (\$\varepsilon 22200, 14200, 10100, 5300), $\lambda_{\rm max}^{\rm MeOH+NaOAc}$ nm 253, 272

sh, 293, 375 (ε 22800, 11400, 9000, 7900), $\lambda_{\text{max}}^{\text{MeOH}+\text{AiCl}_3}$ nm 231, 260, 282, 317 (ε 19100, 17400, 18100, 12100) ¹H NMR (CDCl₃, 270 MHz) of 7-O-acetyl derivative δ 7 90 (s, H-8), 6 95 (s, H-5), 6 90 (dd, J=8, 1 Hz, H-4), 7 57 (dd, J=8 5, 8 Hz, H-3), 6 80 (dd, J=8 5, 1 Hz, H-2), 3 97 (s, OMe), 2 36 (s, OAc) ¹H NMR (CDCl₃, 270 MHz) of 1,7-di-O-acetyl derivative δ 7 88 (s, H-8), 6 93 (s, H-5), 7 37 (dd, J=8, 1 Hz, H-4), 7 67 (dd, J=8 5, 8 Hz, H-3), 7 00 (dd, J=8 5, 1 Hz, H-2), 3 95 (s, OMe), 2 33 and 2 48 (2s, 2 OAc)

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